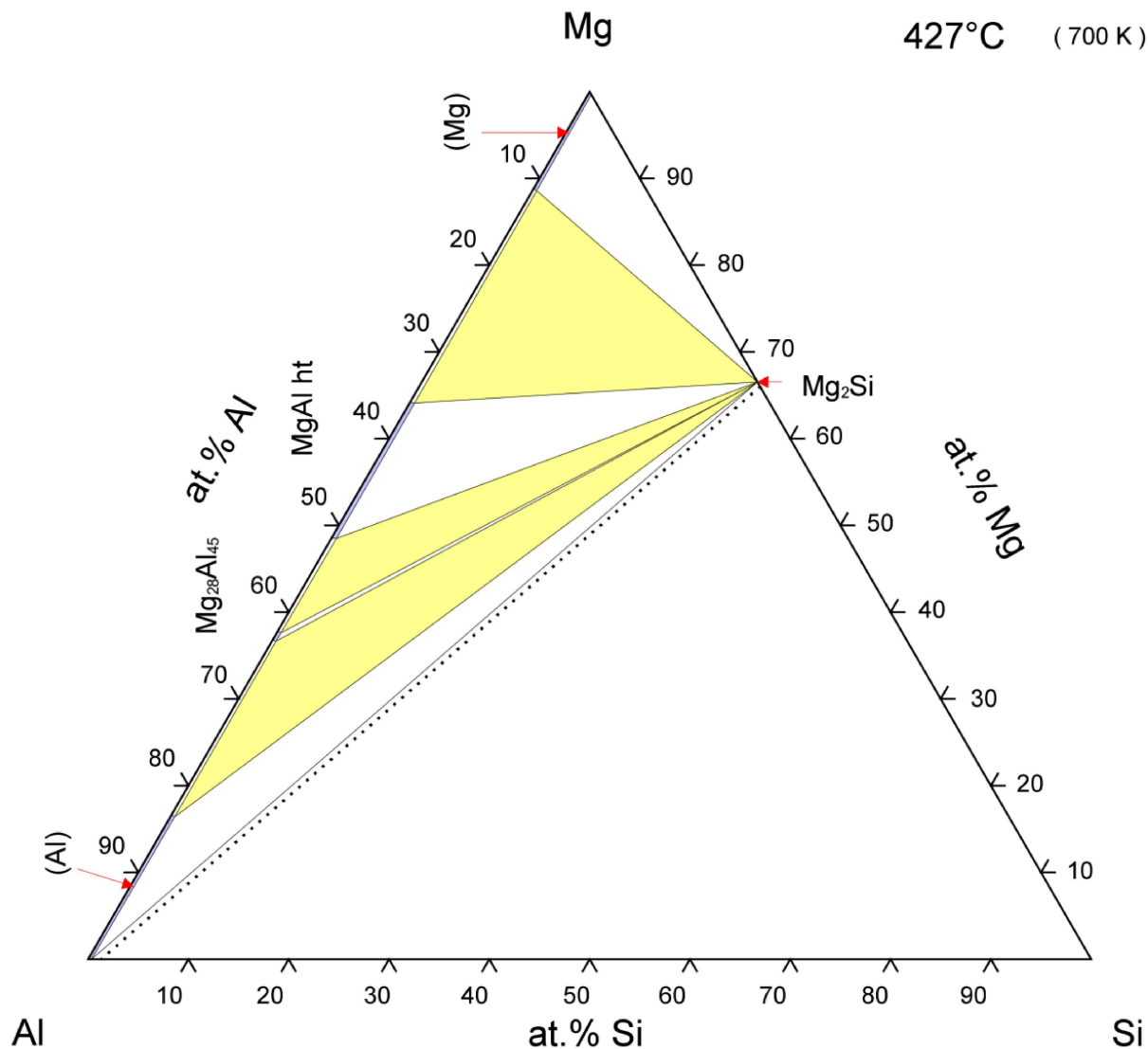


### Aluminum-Magnesium-Silicon Ternary Alloy Phase Diagram (based on 1986 Lüdecke D.)



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<b>Publication Year</b>	1986 [1]
<b>Diagram type</b>	ternary, isothermal section
<b>Concentration range</b>	partial composition; Al-Mg-Mg <sub>2</sub> Si
<b>Temperature</b>	427 °C
<b>Nature of investigation</b>	calculated
<b>APDIC diagram</b>	No

<b>Unique ID No.</b>	976032
<b>Title</b>	Phase Diagram and Thermochemistry of the Al-Mg-Si System
<b>Publication</b>	Z. Metallkd.
<b>Language</b>	English
<b>Authors</b>	Lüdecke D., Aachen RWTH University, Lehrstuhl für Metallurgie der Kernbrennstoffe und Theoretische Hüttenkunde, Aachen, Germany
<b>Original diagram</b>	Al-Mg-Si isothermal section at 427 °C
<b>Original scope</b>	Al conc.[0-100 at.%] vs. Mg conc.[0/66.7-100 at.%] vs. Si conc.[0-33.3 at.%]
<b>Original size</b>	0.7
<b>Remarks</b>	

<b>Crystal data for the complete system</b>							
APD phase label	Formula	Prototype	Pearson symbol	Density Mg/m <sup>3</sup>	Cell parameters		Ref
Published phase label			Space group	Volume nm <sup>3</sup>	nm	°	
Phases shown in this diagram:							
(Al) α	Al	Cu	cF4 Fm-3m	2.7 0.0664	a=0.40497 b=0.40497 c=0.40497 T=295 K	α=90 β=90 γ=90	[2]
(Mg) Mg	Mg	Mg	hP2 P6 <sub>3</sub> /mmc	1.74 0.0466	a=0.32125 b=0.32125 c=0.52132 T=295 K	α=90 β=90 γ=120	[3]
MgAl ht		(no data)					
Mg <sub>28</sub> Al <sub>45</sub> β	Mg <sub>28</sub> Al <sub>45</sub>	Mg <sub>28</sub> Al <sub>45</sub>	cF1832 Fd-3m	2.24 22.5189	a=2.8239 b=2.8239 c=2.8239 T=296(1) K	α=90 β=90 γ=90	[8]
Mg <sub>2</sub> Si Mg <sub>2</sub> Si	Mg <sub>2</sub> Si	CaF <sub>2</sub>	cF12 Fm-3m	1.99 0.25617	a=0.6351 b=0.6351 c=0.6351	α=90 β=90 γ=90	[9]
Phases shown in other Al-Mg-Si diagrams:							
(Si)	Si	C	cF8 Fd-3m	2.33 0.1603	a=0.5432 b=0.5432 c=0.5432	α=90 β=90 γ=90	[4]
Mg <sub>17</sub> Al <sub>12</sub>	Mg <sub>17</sub> Al <sub>12</sub>	Mg <sub>17</sub> Al <sub>12</sub>	cI58 I-43m	2.09 1.1722	a=1.05438 b=1.05438 c=1.05438	α=90 β=90 γ=90	[5]
Mg <sub>23</sub> Al <sub>30</sub> ht	Mg <sub>23</sub> Al <sub>30</sub>	Mg <sub>23</sub> Al <sub>30</sub>	hR159 R-3	2.2 3.09805	a=1.28254 b=1.28254 c=2.17478	α=90 β=90 γ=120	[6]
Mg <sub>1.2</sub> Al <sub>1.8</sub> m1	Mg <sub>1.2</sub> Al <sub>1.8</sub>	MgNi <sub>2</sub>	hP24 P6 <sub>3</sub> /mmc	1.9 0.54252	a=0.573 b=0.573 c=1.908	α=90 β=90 γ=120	[7]
Mg <sub>1.2</sub> Al <sub>1.8</sub> m2	Mg <sub>1.2</sub> Al <sub>1.8</sub>	MgZn <sub>2</sub>	hP12 P6 <sub>3</sub> /mmc	1.9 0.27126	a=0.573 b=0.573 c=0.954	α=90 β=90 γ=120	[7]
MgSi rt		(no data)					

<b>Pure metal melting points and allotropic transformations</b>						
Element phase	Common name	Reaction	Temperature, °C	Prototype	Pearson symbol	Space group
(Al)	αAl	L ↔ (Al)	660.452	Cu	cF4	Fm-3m
(Mg)	Mg	L ↔ (Mg)	650	Mg	hP2	P6 <sub>3</sub> /mmc
(Si)	αSi	L ↔ (Si)	1414	C	cF8	Fd-3m

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For definition of terms and additional phase diagrams information, view the Help pages at [www.asminternational.org/AsmEnterprise/APD](http://www.asminternational.org/AsmEnterprise/APD)

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